

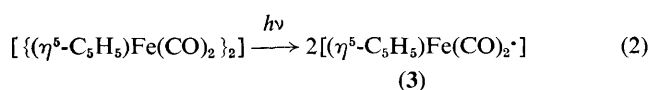
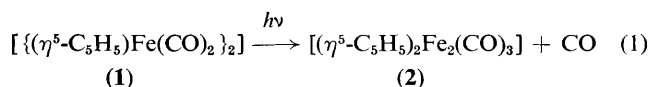
Spectroscopic Evidence for the Formation of Tri- μ -carbonyl-bis-[(η^5 -cyclopentadienyl)iron] on Photolysis of Bis[(dicarbonyl)(η^5 -cyclopentadienyl)iron] in Low Temperature Matrices

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Photolysis of [$\{(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(\text{CO})_2\}_2$] in CH_4 (12 K) and polyvinyl chloride film (12—77 K) matrices leads to the formation of a novel species proposed, on the basis of i.r. evidence including ^{13}C O enrichment and u.v.-visible spectra, to be the CO-bridged dimer [$(\mu\text{-CO})_3\{(\eta^5\text{-C}_5\text{H}_5)\text{Fe}\}_2$].

A recent flash photolysis study¹ of the photochemistry of [$\{(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(\text{CO})_2\}_2$] (**1**) provided evidence for two distinct primary photoprocesses [equations (1) and (2)]. The identities of species (**2**) and (**3**) were proposed on the basis of their lifetimes and u.v.-visible spectra together with the observation of a back reaction of (**2**) with CO; (**2**) was long lived and (**3**) was short lived. No definitive structural evidence was provided.



The photoreactions of (**1**) with phosphine and phosphite ligands² to give substitution products, *e.g.* [$(\eta^5\text{-C}_5\text{H}_5)_2\text{Fe}_2(\text{CO})_3(\text{PPh}_3)$] have been proposed to proceed *via* the CO-bridged intermediate [$(\eta^5\text{-C}_5\text{H}_5)(\text{CO})_2\text{Fe}(\mu\text{-CO})\text{-Fe}(\text{CO})(\eta^5\text{-C}_5\text{H}_5)$]³ and also *via* a radical pathway.⁴

We have studied the photoreactions of (**1**) isolated in polyvinyl chloride films at 12—298 K and in frozen gas matrices at 12 K and we report here the generation of a novel dimeric species which we relate to the intermediates proposed in flash photolysis and solution photoreactions.

The i.r. spectrum of (**1**) in a polyvinyl chloride (pvc) film[†] matrix at 12 K is shown in Figure 1(a). Two terminal CO

[†] The preparation of pvc film matrices and their use over a wide temperature range have been described previously.⁵

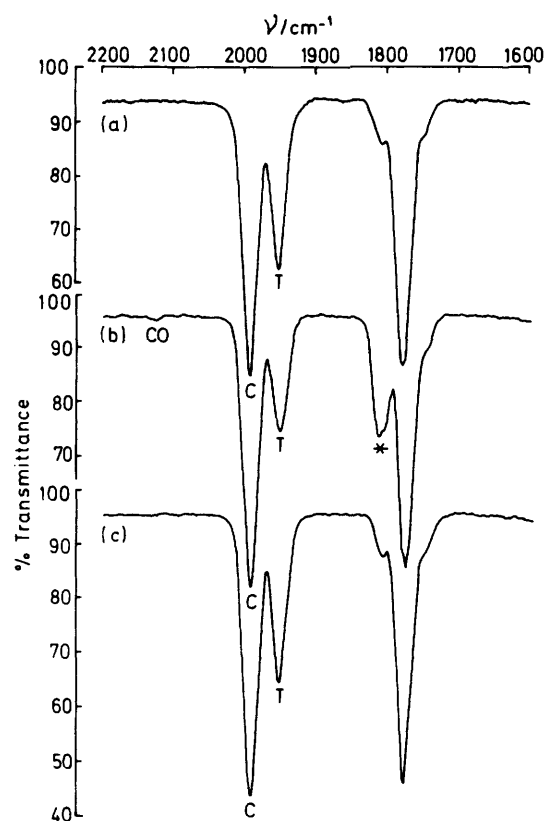


Figure 1. I.r. spectra from an experiment with $[(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(\text{CO})_2]_2$ (**1**) in a pvc film matrix at 12 K: (a) before irradiation, (b) after 150 min irradiation ($\lambda > 400$ nm), and (c) after warming the matrix to 200 K. (Nicolet 7199 FTIR spectrometer; resolution 1 cm^{-1} , 100 scans.) The bands marked C and T are for the *cis*- and *trans*-isomers of (**1**), respectively, and that marked * is for (**4**).

stretching bands at 1992 and 1952 cm^{-1} , which have been assigned to the *cis*- and *trans*-isomers respectively,⁶ are seen together with bridging CO stretching bands at 1804 and 1772 cm^{-1} . On irradiation ($\lambda > 400$ nm) a new absorption appeared at 1812 cm^{-1} together with a band due to free CO at 2133 cm^{-1} whilst the parent bands at 1952 and 1772 cm^{-1} decreased in intensity [Figure 1(b)]. A spectral subtraction showed that there were *no* new product bands in the terminal CO stretching region and that only a slight reduction in the band of the *cis*-isomer of (**1**) had occurred. The photoproduct, which is also observed to be formed at 77 K and in CH_4 matrices at 12 K, has an electronic absorption band exhibiting vibronic fine structure ($\lambda_{\text{max}} = 510\text{ nm}$, $\bar{\nu} = 194\text{ cm}^{-1}$; CH_4 matrix) which is indicative of a highly symmetric species, *cf.* $(\eta^5\text{-C}_5\text{H}_5)_2\text{W}$.⁷ The reaction is reversible so that irradiation with visible light ($\lambda > 500$ nm) or warming a pvc matrix to *ca.* 200 K completely regenerates the parent complex (**1**) [Figure 1(c)].

The photoproduct[‡] is evidently formed by dissociative loss of CO from (**1**) and in particular from the *trans*-isomer exclusively. The fact that it has only one bridging CO stretching band which is i.r. active and this band is at a higher wavenumber than those of (**1**) suggests that the new species is $[(\mu\text{-CO})_3\{(\eta^5\text{-C}_5\text{H}_5)\text{Fe}\}_2]$ with three equivalent bridging CO ligands [(**4**); D_{3h} local symmetry]. The similarly symmetric

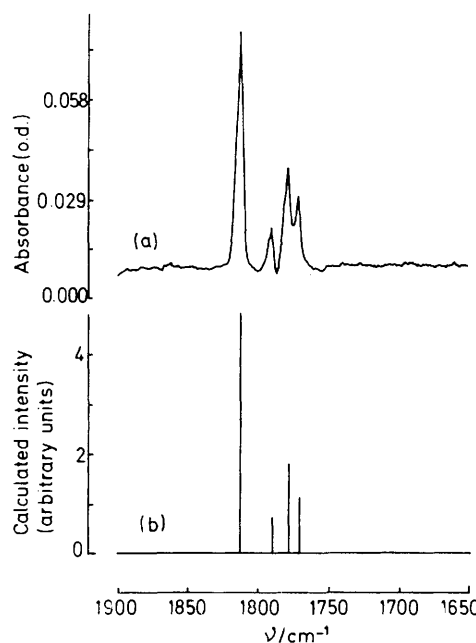
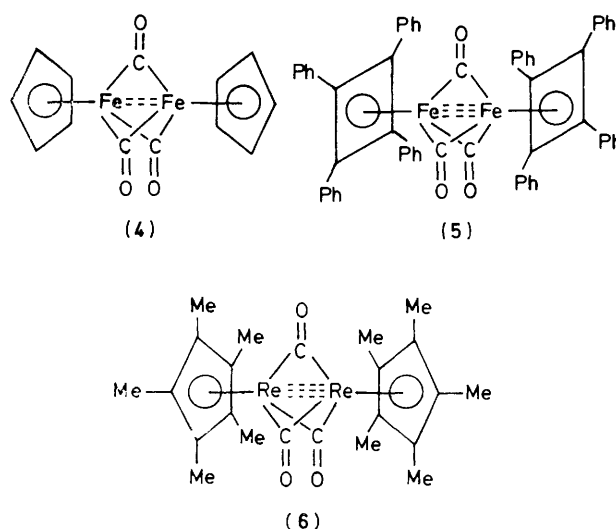


Figure 2. (a) I.r. spectrum (Nicolet 7199 FTIR spectrometer; resolution 1 cm^{-1} , 100 scans) obtained following irradiation ($\lambda > 400$ nm) of a ^{13}CO -enriched (30%) sample of (**1**) in a CH_4 matrix at 12 K and from which the absorptions for the remaining starting material have been subtracted out; (b) calculated spectrum for the CO-bridged species $[\text{M}(^{12}\text{CO})_{3-m}(^{13}\text{CO})_m]$ ($m = 0-3$) where the intensities were obtained for a $^{12}\text{CO}:^{13}\text{CO}$ ratio of 1:0.33 and used the energy-factored CO force constants ($K = 1369.1$ and $k_1 = 41.1\text{ N m}^{-1}$) which gave the best fit between observed and calculated band positions.

dimer $[\text{Fe}_2(\text{CO})_9]$ has one bridging CO stretching band⁹ at 1828 cm^{-1} . The product (**4**) is also similar to the photogenerated dimers $[(\mu\text{-CO})_3\{(\eta^4\text{-C}_6\text{H}_5)\text{Fe}\}_2]$ (**5**) and $[(\mu\text{-CO})_3\{(\eta^5\text{-C}_5\text{Me}_5)\text{Re}\}_2]$ (**6**) which are proposed to have iron-iron¹⁰ and rhenium-rhenium¹¹ triple bonds.

Irradiation ($\lambda > 400$ nm) of $[(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(^{12}\text{CO})_{2-n}(^{13}\text{CO})_n]_2$ ($n = 0-2$)§ in a CH_4 matrix gave new i.r. absorption bands at 1813 , 1791 , 1779 , and 1772 cm^{-1} . The intensity pattern and

‡ Irradiation of (**1**) in pvc films at 12–77 K with u.v. light ($300 < \lambda < 390$ nm) gave rise to another photoproduct which, on warming, reacts with the matrix to form $[(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(\text{CO})_2\text{Cl}]$.⁸

§ Prepared by photolysis ($\lambda > 300$ nm) of (**1**) in hexane-tetrahydrofuran in an atmosphere of ^{13}CO (300 Torr) for 2 h.

positions of these bands could be fitted exactly by using an energy-factored force field for D_{3h} symmetry¹²⁻¹⁴ (Figure 2) and are consistent with the formation of the species $[(\mu\text{-}^{12}\text{CO})_{3-m}(\mu\text{-}^{13}\text{CO})_m\{(\eta^5\text{-C}_5\text{H}_5)\text{Fe}\}_2]$ ($m = 0-3$).

The electronic absorption band of (4) is at the same wavelength (λ_{max} 510 nm) as that of the CO-loss species (2) detected in the flash photolysis experiment.¹ It seems probable, therefore, that (2), which has been proposed to have the stoichiometry shown in equation (1), is identical with (4). Thus the work in low temperature matrices provides the *first* direct evidence for the structure of an intermediate proposed in the photochemical reactions of a metal-metal bonded dimer.

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